Low-frequency Raman spectrum of crystalline 2-(2'-hydroxyphenyl)benzoxazole and density-functional based tight-binding phonon calculations

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Abstract. Unpolarized Raman spectrum of crystalline 2-(2'-hydroxyphenyl) benzoxazole in the region $15-350 \text{ cm}^{-1}$ is compared with the results of density-functional based tight-binding lattice vibration calculations and oriented gas model. The Raman bands observed in the region $25-95 \text{ cm}^{-1}$ are assigned to vibrations that are mainly external; the bands with the frequencies above 150 cm^{-1} are attributed to internal vibrations. It is demonstrated that the molecular in-plane bending vibration, which changes the distance between the oxygen and nitrogen atoms constituting a proton-transfer site, undergoes splitting and mixing with the out-of-plane and external vibrations, giving rise to the Raman bands observed at 115 and 140 cm⁻¹.

Keywords: phonons, molecular crystals, Raman spectra, density-functional based tight-binding method, oriented gas model

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1. Introduction

A phenomenon of photoinduced excited-state intramolecular proton transfer has been attracting much attention of researchers for several decades as an ultrafast photochemical reaction and a process that could have many potential applications, e.g. when tailoring laser dyes and electroluminescent materials [1]. As an example, a photoreactive compound based on excited-state intramolecular proton transfer in a crystalline state has been probed as a possible lasing material [2]. Low-frequency intramolecular vibrations have been considered to play a prominent role in the proton transfer process, changing the distance between a proton donor and an acceptor [3]. In molecular crystals, intermolecular forces give rise to the vibrational motions corresponding to translational and rotational degrees of freedom of molecules. The frequencies of these external vibrations are usually located in the range $0-150 \text{ cm}^{-1}$ and, moreover, new lattice modes treated as a result of mixed internal and external vibrations could appear [4].

According to the X-ray data [5], the crystal lattice of 2-(2^{\circ}-hydroxyphenyl)benzoxazole (HBO) is orthorhombic (Pna2₁), with four molecules per unit cell (see Fig. 1). The adjacent molecules are stacked into chains with interplanar distances 3.58 Å. A high concentration of OH...O rotamers, which corresponds approximately to the ratio 1:1, has been determined by the X-ray data for HBO [5]. However, only the fluorescence band associated with tautomer has been found in crystalline HBO [6]. A study of the infrared spectra of HBO in the region of intramolecular vibrations has not revealed the bands that can be assigned to the vibrations of OH...O rotamers, thus pointing to a low number of this species [7].



(b), and principal axes of inertia of the molecule (a), crystal lattice cell of HBO [5]

The aim of the present work is to elucidate the influence of intermolecular forces on the lowfrequency vibrational spectrum of HBO in its crystalline state.

2. Experimental and computational details

c)

Unpolarized Raman spectrum has been measured at the room temperature in a standard backscattering (135°) geometry, under excitation (532 nm) with the second harmonic of a Q-switched YAG:Nd³⁺ laser LTI-701 that operates at the average power 200 mW. An unpolished single crystal of HBO with the dimensions of $2.0 \times 1.0 \times 0.5$ mm³ is used in our experiments. The spectra have been recorded with a double monochromator DFS-12 equipped with a photomultiplier FEU-79. The spectral resolution is equal to 4 cm⁻¹.

The vibrational modes of HBO crystals have been calculated with a known density-functional based tight-binding method (DFTB), with a third-order correction and a van der Waals interaction correction implemented within a DFTB+ program package [8, 9]. The latter correction was performed using a Lennard–Jones potential with UFF parameters [10]. A $3\times3\times3$ supercell folding has been employed for the Brillouin zone sampling. The phonon calculations have been performed with taking no rotameric disorder into account for the Brillouin zone centre under quasi-harmonic approximation that assumes the cell parameters fixed at the values taken from the X-ray data [5]. The vibrations of the isolated molecule have been calculated in the harmonic approximation. The intensities of the Raman lines have been evaluated within the oriented gas model, using the polarizability and derivatives of the polarizability obtained by density functional theory (DFT) calculations for the isolated molecule. The DFT calculations have been performed basing on a GAMESS program suit [11, 12] and using a B3LYP functional and 6-31G(d,p) basis set.

The motion of a molecule under crystal lattice vibrations can be approximately considered as a translation of its mass centre, a rotation (or libration) and internal vibrations, so that the internal vibrations should satisfy the Eckart conditions [4]. New generalized coordinates for the displacement of molecule k are the centre of mass translation t_{α}^{k} , the angles of rotation about the principal inertia axes θ_{β}^{k} and the internal coordinates $q_{\mu in}^{k}$ ($\mu = 1...3N - 6$, with N being the number of atoms in the molecule). Assuming that the geometry of the molecule in a crystal is close

to the free-molecule geometry, one can take the normal vibrational coordinates $Q_{\mu in}^k$ of the free molecule as generalized internal coordinates. The difference in the geometric parameters causes deviations from the rotational Eckart conditions. Keeping the first-order angular terms only, one can express the Cartesian displacements $\xi_{p\alpha}^k$ of the atom p in the molecule k as [4]

$$\xi_{p\alpha}^{k} = \sum_{\beta=1}^{3} \Lambda_{\alpha\beta}^{k} \left[t_{\beta}^{k} + \sum_{\rho=1}^{3} R_{\beta\rho}^{p} \theta_{\rho}^{k} \right] + \frac{1}{\sqrt{m_{p}}} \sum_{\mu=1}^{3N-6} L_{p\alpha\mu}^{k} Q_{\mu in}^{k} .$$
(1)

Here $\alpha = x, y, z$ is the component index in the coordinate system of crystal, $\beta, \rho = u, v, w$ the component index in the principal inertia molecular axes coordinate system, $L_{\rho\alpha\mu}^{k}$ the eigenvector of the normal vibration of free molecule, R^{p} the skew-symmetry matrix defined as

$$R^{p} = \begin{pmatrix} 0 & w_{p} & -v_{p} \\ -w_{p} & 0 & u_{p} \\ v_{p} & -u_{p} & 0 \end{pmatrix}, u_{p}, v_{p} \text{ and } w_{p} \text{ are the equilibrium atom coordinates in the molecular}$$

coordinate system, $\Lambda_{\alpha\beta}^k$ denotes the matrix of direction cosines of the principal inertia axes of the molecule k and m_p the mass of the atom.

For the Brillouin zone centre ($\mathbf{k} = 0$), the mass-weighted Cartesian and generalized coordinates are related to the phonon normal coordinates Q_i [4]:

$$\sqrt{m_p}\xi_{p\alpha}^k = \sum_j L_{p\alpha j}^k Q_j \,, \, \sqrt{M}t_{\beta}^k = \sum_j L_{\beta j}^k Q_j \,, \, \sqrt{I_{\rho}}\theta_{\rho}^k = \sum_j L_{\rho+3j}^k Q_j \,, \, Q_{\mu in}^k = \sum_j L_{\mu+6j}^k Q_j \,,$$

where $L_{\rho\alpha j}^{k}$ is the component of the eigenvector, *M* the mass of the molecule and I_{ρ} the principal moments of inertia. Using Eq. (1), the $L_{\beta j}^{k}$ components for the generalized coordinates have been obtained.

In the oriented gas approximation, the influence of intermolecular interactions on the optical properties of the molecules and the difference between the macroscopic and local fields are neglected. The Raman tensor of the unit cell in the oriented gas approximation for the normal mode n can be expressed as [4]

$$P_n = \sum_{k=1}^{Z} \frac{\partial \alpha^{(k)}}{\partial Q_n}$$

where $\alpha^{(k)}$ is the molecular polarizability tensor of the molecule k in motionless coordinate system of the crystal and Z the number of molecules in the unit cell:

$$\frac{\partial \alpha^{(k)}}{\partial Q_n} = \sum_{\beta=1}^3 \frac{\partial \alpha^{(k)}}{\partial t_{\beta}^k} \frac{\partial t_{\beta}^k}{\partial Q_n} + \sum_{\rho=1}^3 \frac{\partial \alpha^{(k)}}{\partial \theta_{\rho}^k} \frac{\partial \theta_{\rho}^k}{\partial Q_n} + \sum_{\mu=1}^{3N-6} \frac{\partial \alpha^{(k)}}{\partial Q_{\mu in}^k} \frac{\partial Q_{\mu in}^k}{\partial Q_n} + \frac{\partial \alpha^{(k)}}{\partial \theta_{\rho}^k} \frac{\partial Q_{\mu in}^k}{\partial Q_{\mu in}^k} \frac{\partial Q_{\mu in}^k}{\partial Q_n} + \frac{\partial \alpha^{(k)}}{\partial \theta_{\rho}^k} \frac{\partial \alpha^{(k)}}{\partial \theta_{\rho}^k} = \hat{\Lambda}^k \hat{C}_{\rho} \left(\hat{\Lambda}^k \right)^T.$$

Here $\hat{\Lambda}^k$ is the matrix of direction cosines of the principal inertia axes u, v, w of the molecule k in the equilibrium position,

$$\hat{C}_{u} = \begin{pmatrix} 0 & -\alpha_{uw} & \alpha_{uv} \\ -\alpha_{uw} & -2\alpha_{vw} & \alpha_{vv} - \alpha_{ww} \\ \alpha_{uv} & \alpha_{vv} - \alpha_{ww} & -2\alpha_{vw} \end{pmatrix},$$

$$\hat{C}_{v} = \begin{pmatrix} 2\alpha_{uw} & \alpha_{vw} & \alpha_{wv} - \alpha_{uu} \\ \alpha_{vw} & 0 & -\alpha_{uv} \\ \alpha_{ww} - \alpha_{uu} & -\alpha_{uv} & -2\alpha_{vw} \end{pmatrix},$$

$$\hat{C}_{w} = \begin{pmatrix} -2\alpha_{uv} & \alpha_{uu} - \alpha_{vv} & -\alpha_{vw} \\ \alpha_{uu} - \alpha_{vv} & 2\alpha_{uv} & \alpha_{uw} \\ -\alpha_{vw} & \alpha_{uw} & 0 \end{pmatrix},$$

and $\alpha_{\beta\gamma}$ imply the elements of the molecular polarizability tensor in the coordinate system of the principal inertia axes.

The tensor $\partial \alpha^{(k)} / \partial Q_{\mu in}^k$ is the Raman tensor for the free molecule, which can be expressed in terms of polarizability derivatives with respect to the real displacements [13]:

$$\frac{\partial \alpha^{(k)}}{\partial Q^{k}_{\mu i n}} = \frac{1}{\sqrt{m_{\mu}}} \sum_{n=1}^{3N} \phi^{n}_{\mu} \frac{\partial \alpha^{(k)}}{\partial \xi^{n}},$$

where m_{μ} is the reduced mass of the normal mode, ϕ_{μ}^{n} the normalized real displacements of the atoms in the Cartesian coordinates for the normal mode μ of isolated molecule, $\partial \alpha^{(k)} / \partial \xi^{n}$ the derivative of the polarizability tensor $\alpha^{(k)}$ of the molecule with respect to the real displacements ξ^{n} of atoms, and N the number of atoms in the molecule.

The intensity of the Raman lines can be evaluated according to the formula

$$I_{k\gamma\beta} = \frac{(\omega_{exc} - \omega_k)^4}{\omega_k} [n(\omega_k) + 1] (R_{k\gamma\beta})^2,$$

where ω_{exc} is the laser frequency, $n(\omega_k) + 1 = \left(1 - \exp\left(-\frac{\hbar\omega_k}{k_BT}\right)\right)^{-1}$ the Bose occupation

number, ω_k the frequency of the mode k, k_B the Bolzmann constant, \hbar the Planck constant, T the temperature, and $R_{k\gamma\beta}$ the element of the Raman tensor for the mode k. Here the polarization of the scattered light is directed along the axis γ and the incident radiation is polarized along β . Then the intensity has to been averaged as for polycrystalline samples [14].

We remind that the DFTB is an approximate quantum-chemical method based on the secondorder expansion of a Kohn–Sham total energy in the DFT with respect to the charge-density fluctuations. The unscaled frequencies of vibrations of the isolated HBO molecule calculated with the DFTB and DFT methods are compared in Table 1. The free HBO molecule is planar (the C_s symmetry) and exhibits 47 vibrations with the symmetry A' and 22 vibrations with the symmetry A"; only 9 modes have the frequencies less than 350 cm⁻¹. The next vibration is calculated to have the frequency 407 cm⁻¹. Our results show that the unscaled frequency values obtained with the DFTB method agree well with the experimental parameters for the in-plane molecular vibrations in the region below 350 cm⁻¹, although these frequencies are systematically (within 15 cm⁻¹) lower than the values obtained using the DFT calculations.

	Calc	ulated param	eter	Description	Exp. **	
	DFTB frequency	DFT frequency	DFT Raman	In-plane	Out-of-plane	Frequency
1	17	61			twist	
2	60	67	0.05		out-of-plane bending	
3	112	119	0.94	δ(benzoxazole- CCO)+δ(benzoxazole -CCN)		121
4	151	163	2.60		$\rho(\text{phO-10b})+$ $\rho(\text{ph-10a})$	
5	238	249	0.87		$\rho(\text{phO-10b})+$ $\rho(\text{ph-10b})$	
6	257	266	0.01		ρ(ph-10b)	
7	277	279	2.31	δ(phO-CC)+δ(phO- CCCring)+ δ(benzoxazole- CCO)+δ(benzoxazole -CCN)		280
8	311	317	2.56	$\delta(\text{phO-CCC})+\delta(\text{ph-9b})$		310
9	326	340	0.12		ρ(ph-10a)+ ρ(phO-16b)	

Table 1. Frequencies (in cm^{-1}) and Raman activities (in arbitrary units) of the vibrations associated with isolated HBO molecule in the region below 350 cm⁻¹.

* This description is approximate and uses a Varsanyi's notation [15] for the vibrations of aromatic compounds.

** According to the resonance Raman measurements in solution [3].

3. Results and discussion

As seen from Fig. 2, the unpolarized Raman spectrum for the HBO single crystals in the region 15–350 cm⁻¹ consists of prominent narrow bands located at 33 and 40 cm⁻¹, with a shoulder at 46 cm⁻¹, five overlapping bands with approximately equal intensities in the region 50–130 cm⁻¹, broad bands at 140, 185, 195 and 262 cm⁻¹, and weak narrow bands located at 300 and 316 cm⁻¹. On the other hand, the DFTB phonon calculations predict 57 lattice vibrations (see Table 2) in the frequency region under study. Although these calculations employ no symmetry considerations, the eigenvectors obtained in this way satisfy approximately the symmetry relations. The symmetry types of the vibrations are defined according to the transformations of eigenvectors by the symmetry operations. The eigenvector components $L_{\beta j}^k$ corresponding to the generalized coordinates $\sqrt{M} t_{\alpha}^k$, $\sqrt{I_{\rho}} \theta_{\rho}^k$ and $Q_{\mu in}^k$ are presented in Table 2 for the one molecule. The components for the translational and librational motions do not exceed 0.05 for the vibrations with the frequencies above 150 cm⁻¹, so that the vibrations can be considered as entirely internal.

The results of comparison of the simulated and experimental Raman spectra (see Fig. 2) allow one to assign the bands observed at 185 and 195 cm⁻¹ to the Davydov components A₁ (the calculated frequency 183 cm⁻¹) and A₂ (185 cm⁻¹) of the out-of-plane molecular vibrations, which can be approximately described as $\rho(phO-10b) + \rho(ph-10a)$ according to the Varsanyi's notation [15] (see mode 4 in Table 1). The frequencies calculated for the Davydov quartet are equal to 176, 183, 185 and 189 cm⁻¹, although the components 176 and 189 cm⁻¹ are predicted to exhibit low



Fig. 2. Unpolarized Raman spectrum of HBO single crystals. The calculated frequencies and the corresponding relative peak intensities are represented as full vertical lines.

Table 2. Frequencies ω_i (in cm⁻¹), symmetries *S*, components of the eigenvectors (corresponding to translations along the crystal axes *a*, *b* and *c*, and rotations about the principal molecular axes *u*, *v* and *w*), free-molecule vibrational normal modes (1–9), Raman intensities I_R (in arbitrary units) of the phonon normal modes calculated for the HBO crystals, and frequencies of the Raman bands observed in experiments (ω_{obs}). Generalized coordinates with the components less than 0.05 are not shown.

N	1	2	3	4	5	6	7	8	9	10
ω_i	15	18	29	32	35	37	44	46	48	48
S	B_2	A_2	A_1	B_1	A_2	A_1	A_2	A_1	\mathbf{B}_1	B_2
а	-0.01	-0.04	0.31	0.01	0.00	-0.04	-0.23	0.21	0.00	-0.45
b	0.00	0.05	-0.32	0.45	-0.03	-0.21	0.43	0.11	0.06	0.00
С	0.24	0.33	0.00	0.05	-0.20	0.00	-0.02	0.00	-0.10	-0.07
и	0.43	-0.22	-0.10	-0.05	0.21	0.39	0.07	0.27	0.36	0.02
v	-0.03	0.00	-0.05	0.17	0.03	0.23	0.06	-0.34	0.05	-0.16
W	0.12	-0.30	0.19	-0.01	-0.41	-0.04	0.03	-0.03	-0.11	-0.01
1	0.05	0.05	0.08	-0.07	-0.09	-0.05	-0.04	-0.12	0.30	0.09
2	0.02	0.00	-0.03	0.09	-0.07	-0.07	-0.03	0.04	0.08	-0.08
I_R	0.49	0.00	0.40	0.15	0.01	2.36	0.06	1.43	1.62	0.01
ω_{obs}						33		40	40	
Ν	11	12	13	14	15	16	17	18	19	20
N ω_i	11 50	12 60	13 61	14 64	15 65	16 73	17 74	18 75	19 80	20 82
N ω_i S	11 50 A ₁	12 60 B ₂	13 61 B ₁	14 64 A ₂	15 65 A ₂	16 73 B ₁	17 74 B ₂	18 75 A ₁	19 80 A ₂	20 82 B ₁
N ω_i S a	11 50 A ₁ 0.12	12 60 B ₂ -0.11	13 61 B ₁ 0.00	14 64 A ₂ 0.14	15 65 A ₂ 0.24	16 73 B ₁ 0.00	17 74 B ₂ 0.14	18 75 A ₁ -0.29	19 80 A ₂ 0.25	20 82 B ₁ 0.00
$ \begin{array}{c} N \\ \omega_i \\ S \\ a \\ b \end{array} $	11 50 A ₁ -0.12 -0.11	12 60 B ₂ -0.11 0.00	13 61 B ₁ 0.00 -0.02	14 64 A ₂ 0.14 0.03	15 65 A ₂ 0.24 0.20	16 73 B ₁ 0.00 -0.14	17 74 B ₂ 0.14 0.00	18 75 A ₁ -0.29 -0.25	19 80 A ₂ 0.25 0.09	20 82 B ₁ 0.00 0.01
$ \begin{array}{c} N \\ \omega_i \\ S \\ a \\ b \\ c \end{array} $	11 50 A ₁ -0.12 -0.11 0.00	12 60 B ₂ -0.11 0.00 0.36	$ \begin{array}{c} 13 \\ 61 \\ B_1 \\ \hline 0.00 \\ -0.02 \\ -0.21 \\ \end{array} $	14 64 A ₂ 0.14 0.03 0.29	$ \begin{array}{c} 15 \\ 65 \\ A_2 \\ 0.24 \\ 0.20 \\ -0.12 \end{array} $	$ \begin{array}{c} 16 \\ 73 \\ B_1 \\ 0.00 \\ -0.14 \\ 0.15 \end{array} $	$ \begin{array}{r} 17 \\ 74 \\ B_2 \\ 0.14 \\ 0.00 \\ 0.00 \\ 0.00 \\ \end{array} $	18 75 A ₁ -0.29 -0.25 0.00	19 80 A ₂ 0.25 0.09 -0.02	20 82 B ₁ 0.00 0.01 0.41
N ω _i S a b c u	11 50 A ₁ 0.12 -0.11 0.00 -0.13	12 60 B ₂ -0.11 0.00 0.36 -0.20	$ \begin{array}{c} 13\\ 61\\ B_1\\ \hline 0.00\\ -0.02\\ -0.21\\ -0.29\\ \end{array} $	$ \begin{array}{c} 14 \\ 64 \\ A_2 \\ 0.14 \\ 0.03 \\ 0.29 \\ 0.34 \\ \end{array} $	15 65 A ₂ 0.24 0.20 -0.12 -0.11	$ \begin{array}{c} 16 \\ 73 \\ B_1 \\ \hline 0.00 \\ -0.14 \\ 0.15 \\ 0.07 \\ \end{array} $	$\begin{array}{c} 17 \\ 74 \\ B_2 \\ 0.14 \\ 0.00 \\ 0.00 \\ -0.06 \end{array}$	$ \begin{array}{r} 18\\ 75\\ A_1\\ -0.29\\ -0.25\\ 0.00\\ 0.02\\ \end{array} $	19 80 A ₂ 0.25 0.09 -0.02 -0.15	$\begin{array}{c} 20 \\ 82 \\ B_1 \\ \hline 0.00 \\ 0.01 \\ 0.41 \\ -0.06 \end{array}$
$ \begin{array}{c} N\\ \omega_i\\ S\\ a\\ b\\ c\\ u\\ v\\ \end{array} $	$\begin{array}{c} 11 \\ 50 \\ A_1 \\ \hline 0.12 \\ -0.11 \\ 0.00 \\ -0.13 \\ 0.02 \end{array}$	$\begin{array}{c} 12 \\ 60 \\ B_2 \\ \hline -0.11 \\ 0.00 \\ 0.36 \\ -0.20 \\ 0.04 \end{array}$	$\begin{array}{c} 13 \\ 61 \\ B_1 \\ \hline 0.00 \\ -0.02 \\ -0.21 \\ -0.29 \\ 0.11 \end{array}$	$\begin{array}{c} 14 \\ 64 \\ A_2 \\ \hline 0.14 \\ 0.03 \\ 0.29 \\ 0.34 \\ -0.06 \end{array}$	$ \begin{array}{c} 15\\ 65\\ A_2\\ \hline 0.24\\ 0.20\\ -0.12\\ -0.11\\ -0.28\\ \end{array} $	$\begin{array}{c} 16 \\ 73 \\ B_1 \\ \hline 0.00 \\ -0.14 \\ 0.15 \\ 0.07 \\ 0.44 \end{array}$	$\begin{array}{c} 17 \\ 74 \\ B_2 \\ \hline 0.14 \\ 0.00 \\ 0.00 \\ -0.06 \\ -0.45 \end{array}$	$ \begin{array}{r} 18\\ 75\\ A_1\\ -0.29\\ -0.25\\ 0.00\\ 0.02\\ -0.29\\ \end{array} $	$ \begin{array}{r} 19 \\ 80 \\ A_2 \\ 0.25 \\ 0.09 \\ -0.02 \\ -0.15 \\ 0.32 \\ \end{array} $	20 82 B ₁ 0.00 0.01 0.41 -0.06 -0.11
$ \begin{array}{c} N \\ \omega_i \\ S \\ a \\ b \\ c \\ u \\ v \\ w \end{array} $	$\begin{array}{c} 11\\ 50\\ A_1\\ \hline 0.12\\ -0.11\\ 0.00\\ -0.13\\ 0.02\\ -0.45\\ \end{array}$	$ \begin{array}{c} 12\\ 60\\ B_2\\ \hline -0.11\\ 0.00\\ 0.36\\ -0.20\\ 0.04\\ 0.07\\ \end{array} $	$\begin{array}{c} 13 \\ 61 \\ B_1 \\ \hline 0.00 \\ -0.02 \\ -0.21 \\ -0.29 \\ 0.11 \\ -0.30 \end{array}$	$\begin{array}{c} 14 \\ 64 \\ A_2 \\ \hline 0.14 \\ 0.03 \\ 0.29 \\ 0.34 \\ -0.06 \\ 0.05 \\ \end{array}$	$ \begin{array}{r} 15\\ 65\\ A_2\\ \hline 0.24\\ 0.20\\ -0.12\\ -0.11\\ -0.28\\ -0.04\\ \end{array} $	$\begin{array}{c} 16 \\ 73 \\ B_1 \\ \hline 0.00 \\ -0.14 \\ 0.15 \\ 0.07 \\ 0.44 \\ -0.02 \end{array}$	$\begin{array}{c} 17 \\ 74 \\ B_2 \\ \hline 0.14 \\ 0.00 \\ 0.00 \\ -0.06 \\ -0.45 \\ 0.16 \\ \end{array}$	$ \begin{array}{r} 18\\ 75\\ A_1\\ -0.29\\ -0.25\\ 0.00\\ 0.02\\ -0.29\\ -0.06\\ \end{array} $	$ \begin{array}{r} 19 \\ 80 \\ A_2 \\ 0.25 \\ 0.09 \\ -0.02 \\ -0.15 \\ 0.32 \\ 0.01 \\ \end{array} $	20 82 B ₁ 0.00 0.01 0.41 -0.06 -0.11 -0.18
N ω _i S a b c u v w 1	$\begin{array}{c} 11\\ 50\\ A_1\\ \hline 0.12\\ -0.11\\ 0.00\\ -0.13\\ 0.02\\ -0.45\\ -0.10\\ \end{array}$	$\begin{array}{c} 12 \\ 60 \\ B_2 \\ \hline -0.11 \\ 0.00 \\ 0.36 \\ -0.20 \\ 0.04 \\ 0.07 \\ -0.23 \end{array}$	$\begin{array}{c} 13 \\ 61 \\ B_1 \\ \hline 0.00 \\ -0.02 \\ -0.21 \\ -0.29 \\ 0.11 \\ -0.30 \\ 0.16 \end{array}$	$\begin{array}{c} 14 \\ 64 \\ A_2 \\ \hline 0.14 \\ 0.03 \\ 0.29 \\ 0.34 \\ -0.06 \\ 0.05 \\ -0.16 \\ \end{array}$	$\begin{array}{c} 15 \\ 65 \\ A_2 \\ \hline 0.24 \\ 0.20 \\ -0.12 \\ -0.11 \\ -0.28 \\ -0.04 \\ -0.05 \end{array}$	$\begin{array}{c} 16 \\ 73 \\ B_1 \\ \hline 0.00 \\ -0.14 \\ 0.15 \\ 0.07 \\ 0.44 \\ -0.02 \\ -0.03 \\ \end{array}$	$\begin{array}{c} 17\\ 74\\ B_2\\ \hline 0.14\\ 0.00\\ 0.00\\ -0.06\\ -0.45\\ 0.16\\ -0.05\\ \end{array}$	$ \begin{array}{r} 18\\ 75\\ A_1\\ -0.29\\ -0.25\\ 0.00\\ 0.02\\ -0.29\\ -0.06\\ 0.13\\ \end{array} $	$\begin{array}{c} 19\\ 80\\ A_2\\ \hline 0.25\\ 0.09\\ -0.02\\ -0.15\\ 0.32\\ 0.01\\ -0.21\\ \end{array}$	$\begin{array}{c} 20\\ 82\\ B_1\\ \hline 0.00\\ 0.01\\ 0.41\\ -0.06\\ -0.11\\ -0.18\\ 0.15\\ \end{array}$
$ \begin{array}{c} N \\ \omega_i \\ S \\ a \\ b \\ c \\ u \\ v \\ w \\ 1 \\ 2 \end{array} $	$\begin{array}{c} 11\\ 50\\ A_1\\ \hline 0.12\\ -0.11\\ 0.00\\ -0.13\\ 0.02\\ -0.45\\ -0.10\\ -0.10\\ \hline \end{array}$	$\begin{array}{c} 12 \\ 60 \\ B_2 \\ \hline -0.11 \\ 0.00 \\ 0.36 \\ -0.20 \\ 0.04 \\ 0.07 \\ -0.23 \\ -0.07 \\ \end{array}$	$\begin{array}{c} 13 \\ 61 \\ B_1 \\ \hline 0.00 \\ -0.02 \\ -0.21 \\ -0.29 \\ 0.11 \\ -0.30 \\ 0.16 \\ -0.07 \end{array}$	$\begin{array}{c} 14 \\ 64 \\ A_2 \\ \hline 0.14 \\ 0.03 \\ 0.29 \\ 0.34 \\ -0.06 \\ 0.05 \\ -0.16 \\ 0.06 \\ \end{array}$	$\begin{array}{c} 15 \\ 65 \\ A_2 \\ \hline 0.24 \\ 0.20 \\ -0.12 \\ -0.11 \\ -0.28 \\ -0.04 \\ -0.05 \\ 0.20 \\ \end{array}$	$\begin{array}{c} 16 \\ 73 \\ B_1 \\ \hline 0.00 \\ -0.14 \\ 0.15 \\ 0.07 \\ 0.44 \\ -0.02 \\ -0.03 \\ -0.12 \end{array}$	$\begin{array}{c} 17\\ 74\\ B_2\\ \hline 0.14\\ 0.00\\ 0.00\\ -0.06\\ -0.45\\ 0.16\\ -0.05\\ 0.02\\ \end{array}$	$\begin{array}{c} 18 \\ 75 \\ A_1 \\ -0.29 \\ -0.25 \\ 0.00 \\ 0.02 \\ -0.29 \\ -0.06 \\ 0.13 \\ 0.00 \end{array}$	$\begin{array}{c} 19\\ 80\\ A_2\\ \hline 0.25\\ 0.09\\ -0.02\\ -0.15\\ 0.32\\ 0.01\\ -0.21\\ -0.08\\ \end{array}$	$\begin{array}{c} 20\\ 82\\ B_1\\ \hline 0.00\\ 0.01\\ 0.41\\ -0.06\\ -0.11\\ -0.18\\ 0.15\\ 0.00\\ \end{array}$
N ω_i S a b c u v w 1 2 3	$\begin{array}{c} 11\\ 50\\ A_1\\ \hline 0.12\\ -0.11\\ 0.00\\ -0.13\\ 0.02\\ -0.45\\ -0.10\\ -0.10\\ 0.02\\ \end{array}$	$\begin{array}{c} 12 \\ 60 \\ B_2 \\ \hline -0.11 \\ 0.00 \\ 0.36 \\ -0.20 \\ 0.04 \\ 0.07 \\ -0.23 \\ -0.07 \\ -0.04 \\ \end{array}$	$\begin{array}{c} 13 \\ 61 \\ B_1 \\ \hline 0.00 \\ -0.02 \\ -0.21 \\ -0.29 \\ 0.11 \\ -0.30 \\ 0.16 \\ -0.07 \\ -0.01 \end{array}$	$\begin{array}{c} 14 \\ 64 \\ A_2 \\ \hline 0.14 \\ 0.03 \\ 0.29 \\ 0.34 \\ -0.06 \\ 0.05 \\ -0.16 \\ 0.06 \\ 0.02 \\ \end{array}$	$\begin{array}{c} 15 \\ 65 \\ A_2 \\ \hline 0.24 \\ 0.20 \\ -0.12 \\ -0.11 \\ -0.28 \\ -0.04 \\ -0.05 \\ 0.20 \\ -0.02 \\ \end{array}$	$\begin{array}{c} 16 \\ 73 \\ B_1 \\ \hline 0.00 \\ -0.14 \\ 0.15 \\ 0.07 \\ 0.44 \\ -0.02 \\ -0.03 \\ -0.12 \\ 0.03 \end{array}$	$\begin{array}{c} 17\\ 74\\ B_2\\ \hline 0.14\\ 0.00\\ 0.00\\ -0.06\\ -0.45\\ 0.16\\ -0.05\\ 0.02\\ -0.08\\ \end{array}$	$\begin{array}{c} 18 \\ 75 \\ A_1 \\ \hline -0.29 \\ -0.25 \\ 0.00 \\ 0.02 \\ -0.29 \\ -0.06 \\ 0.13 \\ 0.00 \\ 0.00 \\ \end{array}$	$\begin{array}{c} 19\\ 80\\ A_2\\ \hline 0.25\\ 0.09\\ -0.02\\ -0.15\\ 0.32\\ 0.01\\ -0.21\\ -0.08\\ 0.05\\ \end{array}$	$\begin{array}{c} 20\\82\\B_1\\\hline\\0.00\\0.01\\0.41\\-0.06\\-0.11\\-0.18\\0.15\\0.00\\-0.10\\\end{array}$
N ω_i S a b c u v W 1 2 3 I_R	$\begin{array}{c} 11\\ 50\\ A_1\\ \hline 0.12\\ -0.11\\ 0.00\\ -0.13\\ 0.02\\ -0.45\\ -0.10\\ -0.10\\ 0.02\\ 0.10\\ \end{array}$	$\begin{array}{c} 12 \\ 60 \\ B_2 \\ \hline -0.11 \\ 0.00 \\ 0.36 \\ -0.20 \\ 0.04 \\ 0.07 \\ -0.23 \\ -0.07 \\ -0.04 \\ 0.22 \end{array}$	$\begin{array}{c} 13 \\ 61 \\ B_1 \\ \hline 0.00 \\ -0.02 \\ -0.21 \\ -0.29 \\ 0.11 \\ -0.30 \\ 0.16 \\ -0.07 \\ -0.01 \\ 2.43 \end{array}$	$\begin{array}{c} 14 \\ 64 \\ A_2 \\ \hline 0.14 \\ 0.03 \\ 0.29 \\ 0.34 \\ -0.06 \\ 0.05 \\ -0.16 \\ 0.06 \\ 0.02 \\ 0.03 \\ \end{array}$	$\begin{array}{c} 15 \\ 65 \\ A_2 \\ \hline 0.24 \\ 0.20 \\ -0.12 \\ -0.11 \\ -0.28 \\ -0.04 \\ -0.05 \\ 0.20 \\ -0.02 \\ 0.47 \\ \end{array}$	$\begin{array}{c} 16 \\ 73 \\ B_1 \\ \hline 0.00 \\ -0.14 \\ 0.15 \\ 0.07 \\ 0.44 \\ -0.02 \\ -0.03 \\ -0.12 \\ 0.03 \\ 0.01 \\ \end{array}$	$\begin{array}{c} 17\\ 74\\ B_2\\ \hline 0.14\\ 0.00\\ 0.00\\ -0.06\\ -0.45\\ 0.16\\ -0.05\\ 0.02\\ -0.08\\ 0.34\\ \end{array}$	$\begin{array}{c} 18\\ 75\\ A_1\\ -0.29\\ -0.25\\ 0.00\\ 0.02\\ -0.29\\ -0.06\\ 0.13\\ 0.00\\ 0.00\\ 0.51\\ \end{array}$	$\begin{array}{c} 19\\ 80\\ A_2\\ \hline 0.25\\ 0.09\\ -0.02\\ -0.15\\ 0.32\\ 0.01\\ -0.21\\ -0.08\\ 0.05\\ 0.60\\ \end{array}$	$\begin{array}{c} 20\\82\\B_1\\\hline\\0.00\\0.01\\0.41\\-0.06\\-0.11\\-0.18\\0.15\\0.00\\-0.10\\0.15\\\end{array}$

Low-frequency Raman

17	1				2.5	2(20	20	20
Ν	21	22	23	24	25	26	27	28	29	30
ω_i	96 D	98	103 D	104	109	109 D	110 D	116	121	123
3	B ₂	A ₂	B ₂	A ₁	A ₂	B ₂	B ₁	A ₁	A ₁	A ₂
a h	-0.05	0.19	-0.10	-0.10	-0.11	0.02	0.00	-0.03	-0.04	-0.04
D C	0.01	0.10	0.00	-0.05	-0.00	-0.01	0.03	-0.09	-0.10	-0.04
<i>u</i>	0.09	0.00	-0.03	-0.00	0.00	0.05	0.05	-0.07	-0.02	0.02
v	0.05	0.15	-0.01	-0.07	0.00	-0.13	-0.08	0.07	0.02	0.02
w	0.29	-0.03	0.00	0.14	-0.03	-0.27	-0.25	0.00	0.01	-0.02
1	-0.29	0.39	-0.06	-0.35	-0.03	-0.16	-0.16	-0.13	-0.25	-0.09
2	-0.05	0.04	0.48	-0.28	0.41	-0.03	-0.09	0.28	0.26	0.12
3	0.10	-0.05	0.03	0.07	0.17	0.37	0.38	0.37	-0.32	-0.46
4	0.03	0.08	0.09	-0.03	0.10	0.01	0.05	0.09	0.03	0.02
I_R	0.36	0.01	0.01	0.07	0.00	0.47	0.02	0.10	0.04	0.09
ω_{obs}	102					115				
N	31	32	33	34	35	36	37	38	39	40
();	126	140	141	176	183	185	189	248	251	256
S	B ₁	Ba	B ₁	B ₁	A,	A	Ba	B 1	A	<u>A</u> 1
	0.00	-0.02	0.00	0.00	_0.03	0.01	0.02	0.00	001	0.01
h h	-0.12	0.02	0.00	0.00	0.00	-0.01	0.02	0.00	-0.01	0.01
c	0.04	-0.08	0.06	-0.01	0.00	0.01	0.00	-0.01	0.00	0.00
и	-0.09	0.15	-0.14	0.02	0.01	-0.01	0.00	-0.01	-0.01	0.01
ν	0.07	-0.01	0.00	-0.03	-0.04	-0.10	-0.06	0.02	0.03	0.00
w	-0.01	-0.26	0.25	0.05	-0.02	0.02	-0.03	0.02	0.00	0.02
1	0.00	-0.25	0.28	0.02	0.03	-0.05	0.03	0.00	-0.01	-0.02
2	0.43	-0.01	-0.08	-0.14	-0.08	-0.08	-0.08	0.02	0.02	0.00
3	0.12	-0.30	0.28	-0.05	-0.04	-0.02	-0.02	0.00	-0.01	0.01
4	0.16	0.00	-0.05	0.47	0.49	0.48	0.49	-0.04	-0.07	-0.02
5	0.00	-0.02	0.03	-0.04	0.03	-0.07	-0.01	-0.49	-0.48	0.48
6 7	0.00	-0.04	0.05	0.01	-0.01	-0.03	-0.02	0.06	-0.03	0.10
8	0.01	0.01	-0.02	0.00	0.01	-0.02	-0.04	0.02	0.10	-0.05
9	-0.02	0.00	0.01	0.01	0.00	0.00	0.00	-0.03	0.04	-0.03
I_{P}	0.02	0.00	0.00	0.00	0.01	0.02	0.00	0.00	0.00	0.01
ω_{obs}	0.00	140	0.00	0.00	185	195	0.01	0.00	0.01	262
N	41	42	43	44	45	46	47	48	49	50
ω_i	258	269	269	270	271	280	281	284	285	314
S	B_2	B_1	A ₂	A_1	B ₂	B_2	A ₂	B_1	A ₁	A_1
5	-0.48	0.09	0.06	0.03	0.10	0.06	0.09	-0.01	-0.04	-0.05
6	0.10	0.48	0.48	-0.49	-0.48	-0.03	-0.02	0.07	0.06	-0.03
0	0.06	0.02	0.07	-0.02	-0.06	0.49	0.49	-0.49	-0.49	-0.03
8 0	0.05	-0.08	-0.07	0.04	0.03	-0.01	-0.02	0.01	0.03	-0.49
9 I.	-0.04	0.01	-0.02	0.01	0.04	0.00	-0.01	0.02	0.02	0.02
Ω_{abs}	0.00	0.00	0.00	0.00	0.00	0.00	0.02	0.00	300	316
<i>** 005</i>										
N	51	52	53	54	55	56	57			
ω_i	315	318	320	340	340	342	346			
S	A_2	B_2	B_1	A_2	B_2	B_1	A_1			
5	-0.04	-0.07	-0.06	-0.01	-0.04	0.00	-0.01			
6	-0.04	-0.06	-0.06	-0.02	0.00	-0.02	0.04			
7	-0.02	-0.01	-0.02	-0.01	0.01	-0.02	0.02			
8	-0.49	-0.49	-0.49	-0.08	0.03	-0.01	0.02			
9 In	0.09	0.02	0.01	-0.49	0.50	-0.50	0.49			
M_R	0.01	0.00	0.01	0.00	0.00	0.00	0.01			
w _{obs}										

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intensity. The band observed at 262 cm⁻¹ is assigned to the out-of-plane molecular vibration ρ (ph-10b). The dominant component is determined to have the symmetry A₁ (256 cm⁻¹). The band located at 300 cm⁻¹ is due to the A₁ component (285 cm⁻¹). Finally, the band located at 316 cm⁻¹ corresponds to the molecular bending vibration δ (phO-CCC)+ δ (ph-9b). The largest intensity is obtained for the A₁ component (314 cm⁻¹).

Table 3. Correlations among molecular symmetry, site symmetry and factor-group symmetry of intramolecular vibrations in the crystalline HBO.



The geometry of the HBO molecules calculated inside the crystal is close to planar; the dihedral angles formed by benzoxazole and phenolic moieties do not exceed 3°. Moreover, the isolated molecule is calculated to be more precisely planar. The molecules occupy general positions in the crystal cell, so that accurate separation of the translational and librational motions is not possible and the site symmetry remains C_1 . The dominant translational displacements are predicted for the vibrations with the calculated frequencies 29, 32, 44, 48 (B₂), 60 and 82 cm⁻¹, whereas the rotational displacements are predicted for the vibration frequencies 15, 35, 37, 46, 50, 61, 73 and 74 cm⁻¹. The crystal mode located at 104 cm⁻¹ corresponds mainly to the internal vibrations which can be described as a mixture of the out-of-plane molecular modes at 47 and 60 cm⁻¹ (see Table 2). The mode at 48 cm⁻¹ represents the internal vibration mixed with libration, while the modes located at 96 and 98 $\rm cm^{-1}$ are the internal vibration mixed with both libration and translation. The vibrations at 103 (B₂), 109 (A₂) and 126 cm⁻¹ (B₁) are predominantly internal and correspond to the out-of-plane molecular vibration 60 $\rm cm^{-1}$. The seven modes lying in the region 109–141 cm⁻¹ involve the internal vibration similar to the in-plane molecular mode δ (benzoxazole-CCO)+ δ (benzoxazole-CCN) (the calculated frequency being equal to 112 cm⁻¹ and the experimental one to 121 cm⁻¹ [3]). Three of them are predominantly internal (116 (A₁), 121 (A₁)) and 123 cm^{-1} (A₂)) and the remaining vibrations are mixed with librations. However, the modes 116 and 121 cm⁻¹ are mixtures of the in-plane and out-of-plane deformations that occur owing to symmetry lowering down to C_1 (see Table 3). The in-plane vibration modulates the distance between the O and N atoms. Finally, all the other modes lying in the region below 150 cm⁻¹ represent mainly the mixed translation-librational external vibrations.

The frequencies of the most intense Raman bands obtained following from our calculations demonstrate a good agreement with those of the bands actually observed in the measured spectrum. For instance, the intense band found theoretically at 37 cm⁻¹, which originates from the full-symmetry A₁ librational vibration around the principal axes *u* and *w*, is quite close to the narrow band observed at 33 cm⁻¹. The strong narrow band observed at 40 cm⁻¹ can be attributed to overlapping of the bands caused by the full-symmetry (A₁) librational vibration around the principal axes *u* and *w* (the calculated frequency being 46 cm⁻¹) and the mixed mode B₁ located at 48 cm⁻¹. The shoulder at 46 cm⁻¹ can be assigned to the full-symmetry (A₁) vibration 50 cm⁻¹, which is predominantly libration around the long molecular axis *w*. The band at 62 cm⁻¹

corresponds to the B_1 libration (61 cm⁻¹). The location at 70 cm⁻¹ for one of the bands is close to 65 cm⁻¹, as calculated for the translational-librational vibration A_2 . The wider band at 85 cm⁻¹ can be attributed to the overlapped lines originating from the librational mode (74 cm⁻¹) and translational-librational modes A_1 and A_2 (75 and 80 cm⁻¹, respectively). The bands calculated as the mixed modes located at 96 cm⁻¹ (out-of-plane twisting, translation along the *c* axis and libration about the *w* axis with the symmetry B_2) and at 109 and 140 cm⁻¹ (mixture of in-plane bend, out-of-plane twisting and libration about the axis *w* with the symmetry B_2) agree well with the experimental bands found at 102, 115 and 140 cm⁻¹. In general, the simulated spectrum exhibits a qualitative rather than quantitative agreement with the observed one, thus predicting unambiguously only the frequencies of the most intense bands.

4. Conclusion

The interactions among the HBO molecules in the crystalline state lead to appearance of the lattice vibrations in the region $15-150 \text{ cm}^{-1}$, which is typical for the organic molecular crystals. The lattice vibrations in the frequency region $15-95 \text{ cm}^{-1}$ are mainly external, while a considerable contribution of intramolecular vibrations is found only for the twisting internal vibrations. In the region $95-150 \text{ cm}^{-1}$, the contribution of internal degrees of freedom is more significant. The inplane molecular bending vibration changes the distance between the O and N atoms constituting the hydrogen-transfer site and becomes split and mixed with both the out-of-plane and external vibrations. The vibrations located above 150 cm^{-1} are entirely internal and retain similarity to the corresponding molecular vibrations; a considerable splitting is found for the out-of-plane vibration located at 185 cm^{-1} .

The oriented gas model demonstrates that modulations of polarizability of the crystal originating from librations of the molecules, under the assumption of constant molecular polarizability, can be responsible for the intense bands observed in the spectrum. Finally, we are to note that the oriented gas model represents only a rough estimation of the relative intensities since it does not take into account the influence of intermolecular interactions on the polarizability of molecules.

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Syetov Y. 2017. Low-frequency Raman spectrum of crystalline 2-(2'-hydroxyphenyl)benzoxazole and density-functional based tight-binding phonon calculations. Ukr.J.Phys.Opt. **18**: 67 – 76.

Анотація. Неполяризований спектр комбінаційного розсіяння світла кристалічного 2-(2'-гідроксифеніл) бензоксазолу в області $15-350 \text{ см}^{-1}$ порівняно з результатами розрахунку коливань кристалічної гратки за допомогою методу функціонала густини в наближенні сильного зв'язку та моделі орієнтованого газу. Смуги, що спостерігаються в діапазоні $25-95 \text{ см}^{-1}$, віднесено до коливань, які є переважно зовнішніми, а смуги з частотами, вищими за 150 см^{-1} , віднесено до внутрішніх коливань. Молекулярне площинне деформаційне коливання, що змінює відстань між атомами кисню та азоту, на яких має місце перенесення протона, піддається розщепленню та змішуванню з позаплощинними і зовнішніми коливаннями. Це коливання відповідає смугам у спектрі поблизу 115 і 140 см⁻¹.